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Key indicators

Single-crystal X-ray study
 $T = 273$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.059
 wR factor = 0.140
Data-to-parameter ratio = 16.6For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.*N*-(2-Chlorophenyl)-*N'*-(4-methoxybenzoyl)-
thiourea

In the title compound, $\text{C}_{15}\text{H}_{13}\text{ClN}_2\text{O}_2\text{S}$, the dihedral angle between the 2-chlorophenyl and 4-methoxyphenyl groups is $54.12(13)^\circ$. The molecule is stabilized by intermolecular $\text{C}-\text{H}\cdots\text{S}$ and $\text{N}-\text{H}\cdots\text{S}$ interactions, forming double-column chains arranged parallel to the b axis.

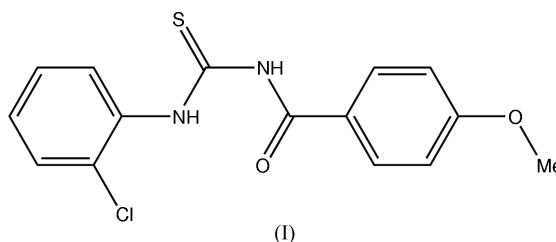
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Comment

The molecular structure and dimensions of the title compound, (I), are similar to those of other benzoylthiourea derivatives, such as *N*-benzoyl-*N'*-(2-chlorophenyl)thiourea (Yusof & Yamin, 2004) and *N*-benzoyl-*N'*-phenylthiourea (Yamin & Yusof, 2003). The molecule maintains its *cis-trans* configuration with respect to the position of the 4-methoxyphenyl and 2-chlorophenyl groups relative to the S atom across the thiourea C–N bonds.



The central carbonylthiourea moiety (S1/C8/N1/N2/C9/O1), 4-methoxyphenyl (C1–C6/O2/C15) and 2-chlorophenyl (C9–C14/Cl1) fragments are each planar, the maximum deviation being $0.075(3)$ Å for atom C15. The central carbonylthiourea moiety makes dihedral angles of $9.91(10)$ and $44.23(11)^\circ$, respectively, with the 4-methoxyphenyl and 2-chlorophenyl fragments. This can be compared with values of $11.96(9)$ and $29.36(8)^\circ$ for the same angles in *N*-benzoyl-*N'*-(2-chlorophenyl)thiourea. However, the dihedral angle between the 4-methoxyphenyl and 2-chlorophenyl fragments of $54.12(13)^\circ$ is

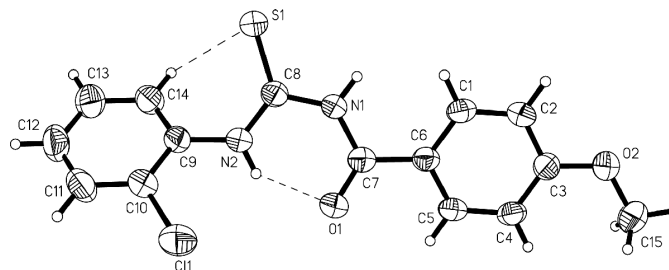


Figure 1

The molecular structure of compound (I), shown with 50% probability displacement ellipsoids. Dashed lines indicate intramolecular hydrogen bonds.

larger than the value of $38.67(10)^\circ$ for *N*-benzoyl-*N'*-(2-chlorophenyl)thiourea.

There are two intramolecular hydrogen bonds, N2—H2···O1 and C14—H14···S1 (Table 2), and as a result, two pseudo-six-membered rings (S1—C8—N2—C9—C14—H14 and O1—C7—N1—C8—N2—H2) are formed. In the crystal structure, the molecules are linked by intermolecular interactions, C1—H1A···S1ⁱ and N1—H1···S1ⁱⁱ [symmetry codes: (i) $\frac{3}{2} - x, y - \frac{1}{2}, \frac{1}{2} - z$; (ii) $\frac{3}{2} - x, \frac{1}{2} + y, \frac{1}{2} - z$; Table 2], forming double-column chains along the *b* axis.

Experimental

A solution of 2-(chlorophenyl)aniline (2.00 g, 0.016 mol) in acetone (50 ml) was added dropwise to an acetone solution (50 ml) containing an equimolar amount of 4-methoxyphenylbenzoyl isothiocyanate in a two-necked round-bottomed flask. The solution was refluxed for about 2 h and then cooled in ice. The white precipitate was filtered off and washed with ethanol—distilled water, then dried in a vacuum (yield 87%). Recrystallization from ethanol yielded single crystals suitable for X-ray analysis.

Crystal data

| | |
|-------------------------------|---|
| $C_{15}H_{13}ClN_2O_2S$ | $D_x = 1.453 \text{ Mg m}^{-3}$ |
| $M_r = 320.78$ | Mo $K\alpha$ radiation |
| Monoclinic, $P2_1/n$ | Cell parameters from 2167 reflections |
| $a = 13.438(3) \text{ \AA}$ | $\theta = 1.5\text{--}27.0^\circ$ |
| $b = 3.9893(9) \text{ \AA}$ | $\mu = 0.41 \text{ mm}^{-1}$ |
| $c = 27.727(6) \text{ \AA}$ | $T = 273(2) \text{ K}$ |
| $\beta = 99.508(4)^\circ$ | Block, colourless |
| $V = 1466.0(5) \text{ \AA}^3$ | $0.22 \times 0.19 \times 0.16 \text{ mm}$ |
| $Z = 4$ | |

Data collection

| | |
|---|--|
| Bruker SMART APEX CCD area-detector diffractometer | 3162 independent reflections |
| ω scans | 2649 reflections with $I > 2\sigma(I)$ |
| Absorption correction: multi-scan (SADABS; Sheldrick, 1996) | $R_{\text{int}} = 0.024$ |
| $T_{\text{min}} = 0.916, T_{\text{max}} = 0.938$ | $\theta_{\text{max}} = 27.0^\circ$ |
| 7879 measured reflections | $h = -13 \rightarrow 17$ |
| | $k = -5 \rightarrow 5$ |
| | $l = -35 \rightarrow 27$ |

Refinement

| | |
|---------------------------------|--|
| Refinement on F^2 | $w = 1/[\sigma^2(F_o^2) + (0.0589P)^2 + 0.3777P]$ |
| $R[F^2 > 2\sigma(F^2)] = 0.059$ | where $P = (F_o^2 + 2F_c^2)/3$ |
| $wR(F^2) = 0.140$ | $(\Delta/\sigma)_{\text{max}} = 0.001$ |
| $S = 1.19$ | $\Delta\rho_{\text{max}} = 0.33 \text{ e \AA}^{-3}$ |
| 3162 reflections | $\Delta\rho_{\text{min}} = -0.18 \text{ e \AA}^{-3}$ |
| 191 parameters | |
| H-atom parameters constrained | |

Table 1

Selected geometric parameters ($\text{\AA}, ^\circ$).

| | | | |
|-------------|-----------|-------------|----------|
| C11—C10 | 1.731(3) | N1—C8 | 1.386(3) |
| S1—C8 | 1.662(2) | N2—C8 | 1.331(3) |
| N1—C7 | 1.377(3) | N2—C9 | 1.415(3) |
| C9—N2—C8—N1 | −179.3(2) | C7—N1—C8—S1 | 177.4(2) |

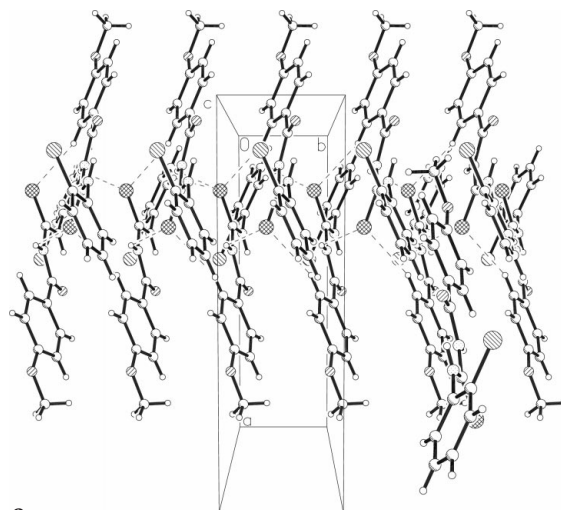


Figure 2

Packing diagram of (I), viewed down the *c* axis. Dashed lines indicate the C—H···S hydrogen bonds.

Table 2

Hydrogen-bonding geometry ($\text{\AA}, ^\circ$).

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|---------------------------|-------|-------------|-------------|---------------|
| N2—H2···O1 | 0.86 | 1.93 | 2.638(3) | 138 |
| C14—H14···S1 | 0.93 | 2.80 | 3.209(3) | 107 |
| C1—H1B···S1 ⁱ | 0.93 | 2.80 | 3.686(3) | 160 |
| N1—H1A···S1 ⁱⁱ | 0.86 | 2.87 | 3.478(3) | 129 |

Symmetry codes: (i) $\frac{3}{2} - x, y - \frac{1}{2}, \frac{1}{2} - z$; (ii) $\frac{3}{2} - x, \frac{1}{2} + y, \frac{1}{2} - z$.

After their location in a difference map, all H atoms were placed geometrically in ideal positions and allowed to ride on the parent atoms, with C—H = 0.93–0.96 \AA and N—H = 0.86 \AA , and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{methyl C})$ or $1.2U_{\text{eq}}(\text{C,N})$.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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